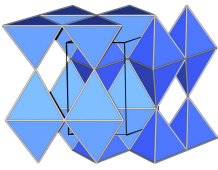
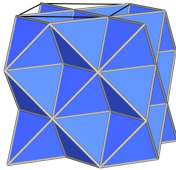
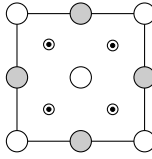
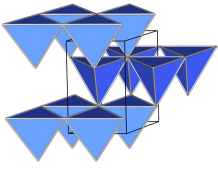
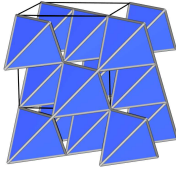
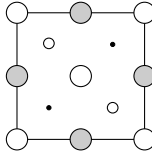
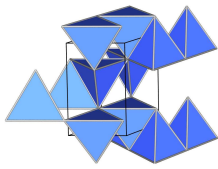
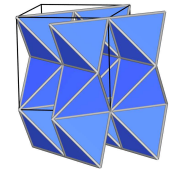
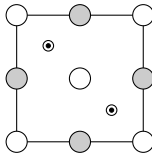
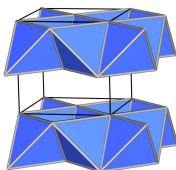
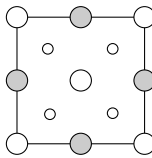
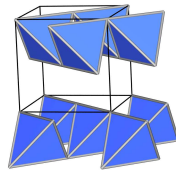
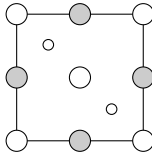
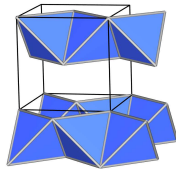
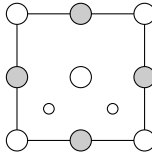
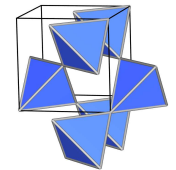
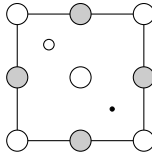


1.1.4. Ionenkristalle (Forts.)

Strukturchemie II: Dichte Packungen mit gefüllten Tetraederlücken

X hexagonal dicht	Formel	CN (X)	Lückenbes.	X kubisch dicht
	M_2X	4:8	1	  <p>Li_2O (anti-CaF_2)</p>
	M_3X_2	4:6	3/4	O_3Mn_2 (C-Str., defekt-anti- CaF_2)
 <p>Wurtzit</p>	MX	4:4	1/2	  <p>Zinkblende</p>
 <p>β-BeO</p>				  <p>PtS (CuO)</p>
				  <p>OPb</p>
	MX_2	4:2	1/4	  <p>HgI_2</p>
β -ZnCl ₂				  <p>SiS_2</p>
				  <p>OCu₂</p>
$SnBr_4$	MX_4	4:1	1/8	OsO_4